

Y POLITECNICO DI MILANO



Giuseppina Raffaini

Dipartimento di Chimica, Materiali e Ing. Chimica "G. Natta" Politecnico di Milano - Italy





Simulation study

of protein adsorption on biomaterials surfaces:

• TiO₂

Carbon Nanotubes (CNT)

comparing our theoretical results with experimental data



- Ti : widely used in medical *implants and prostheses* due to its bulk and surface properties.
- It interacts with biological fluids through a passivating TiO₂ film
 We wish to control the film thickness, morphology and crystal structure
 - Three TiO₂ polymorphs: rutile, anatase and brookite



They expose **unlike surfaces**, which may *differently* **affect protein adsorption** \\$hence *biocompatibility* and *performance* of the biomaterial.





- only bridging oxygen and Ti atoms are exposed
- Similar surface chemistry : in this case we can study the effect of the nanoscale topography only

Protein fragments:

- ✓ ALB = albumin subdomain (α -helices)
- FIB = fibronectin modules (β-sheets)



•A key issue: at a fixed surface chemistry, does the surface topography affects protein adsorption?

Current simulation methods based on molecular mechanics and molecular dynamics can (correctly) predict significant differences?



Molecular studies at atomistic level based on Molecular Mechanics and Molecular Dynamics

Molecular Mechanics

Energy minimization with respect to all the variables (the atomic coordinates) of protein fragments near different surfaces

- Segmetry of interaction (conformational changes)
- \$ strength of interaction (interaction energy E_{int} , strain energy E_{strain})
- It surface coverage (total or partial) and film formation

Molecular Dynamics

Time evolution of the system *at constant (average) T* solving the classical equations of motions (Newton) for each atom

- kinetics of adsorption process (kinetics of spreading)
- mobility on the surface
- > possible surface ordering induced by the surface







Starting with different trial orientations

Solution The energy minimizations yield the most stable *initial* geometries:



G→ Local deformations to enhance the contact surface and *local* loss of secondary structure

Ger Other higher energy geometries are present (local energy minima)

For all initial geometries the E_{int} and E_{strain} are calculated...





For both proteins we find:

✓ More favorable interaction on anatase surface

✓ Brookite shows a significantly weaker interaction

Strain energy (broken H-bonds, ...) increases more slowly than E_{int}
 This behavior implies that larger deformations may enhance the interaction strength...

2. Second step of proposed simulation protocol **MD** : Kinetics of spreading





When the protein fragments spread on the surface:

1. the potential energy decreases smoothly

2. while the distance can show some jumps due to conformational changes

Ser Faster spreading for albumin (a 'soft' protein)

FIB

3. Third step of proposed simulation protocol MM after MD \Rightarrow Final adsorption stage



Most stable final geometries



Surface spreading and flattening: more residues interact with the surface

Stronger interaction with optimization of both:

- protein-surface interactions
- intra-molecular interactions (that can take place at longer time)

For both protein fragments Stronger adsorption on anatase as an effect of the surface topography!

On anatase, stronger interaction with FIB that keeps a secondary structure similar to the native one

- Considering the intrinsic *E*_{int} and the conformation, the fibronectin fragments:
 - 1. show stronger interaction with anatase surface than albumin

in agreement with competitive adsorption experiments HSA and Fn on TiO₂

HSA adsorption is faster (larger diffusivity) but is then replaced by Fn

SR Sousa et al. J. Biomed. Mat. Res. A, 84, 281-290, 2008

2. Retain their secondary and tertiary structure

then, their functionality, that is to mediate the cell adhesion
 then, important for osteoblast adhesion and osteointegration

in agreement with experiments on TiO₂ polymorphs Anatase shows better osteoblast adhesion (driven by Fn adsorption!) and osteointegration than rutile

R Chiesa et al., *J Appl Biomater Biomech*, 1, 91-107, 2003 R Chiesa et al., *Oral Surg Oral Med Oral Pathol Oral Radiol Endod*, 103, 745-756, 2007

.. about Protein Adsorption on

CNT surfaces

C allotropes

substrates with same surface chemistry but different curvature, then topography

using the same methodology

comparing theoretical results with experimental data

Solution States Sta

✤ Then, a larger curvature yields a weaker interaction!

Still, it is stronger than on Hydrophilic PVA amorphous surface

... Results reported in my PhD Thesis In Materials Engineering at the Politecnico of Milan

The energetic cost to detach a CNT from a random aggregate

IInd result: Final adsorption stage

is less than the energy gain due to adsorption

(calculated after MM and MD runs in the most stable adsorption geometry)

\Rightarrow hence we find that CNT can be solubilized in water by proteins through non covalent interactions

... AS INDEED EXPERIMENTALLY FOUND JS Dordick, RS Kane et al

Langmuir, 22, 1392-1395 (2006)

on OUTER CNT surface

BSA

water

MJL

... Finally, recent results about <u>competitive</u> adsorption:

on TiO₂ anatase nanocrystal surface

• of <u>small</u> molecules (quinoline molecules)...

Anatase TiO₂ can exposed different crystal surfaces

.... Considering ideal TiO₂ anatase nanocrystal:

Anatase (001) hydroxylated surface

Anatase (00-1) Titanium atoms exposed

Anatase (100) bridging oxygens

RSC Publishing

- MM and MD simulations are most useful to study protein adsorption: with atomistic details ⇒ we can study the formation, structure and stability of the physisorbed layer
- We can model the effect of the nanoscale topography (roughness, curvature, ...).
- The interaction strength is related also to the protein size and rigidity, and the nature of aminoacids in contact with the specific surface (hydrophilic or hydrophobic).
- TiO₂ polymorphs: anatase leads to a stronger adsorption than rutile, in particular for fibronectin which also preserves the native structure and functionality.
- CNT: strong adsorption dependent on the curvature (concavity or convexity)
- proteins can solubilize CNT in water through non-covalent interactions
- concave surfaces of appropriate curvature provide a stronger adsorption
- Ordered surfaces may induce an intramolecular parallel arrangement.
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